

0040-4020(95)01079-3

Terpenoids with Antifouling Activity against Barnacle Larvae from the Marine Sponge *Acanthella cavernosa*

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Abstract: Fourteen terpenoids, including six new compounds, have been isolated from the marine sponge Acanthella cavernosa collected off Hachijo-jima Island. They inhibit of larval attachment and metamorphosis of the barnacle Balanus amphitrite. Their structures were determined mainly by spectroscopic methods.

INTRODUCTION

Sessile marine organisms possess various defense systems against predators, against settling larvae of other sessile organisms, and against pathogenic microbes. Hence chemical defense is significant in these organisms and their secondary metabolites are potential antifouling agents.¹

During our screening for antifouling compounds from Japanese marine invertebrates,² the marine sponge *Acanthella cavernosa* collected off Hachijo-jima Island (300 km south of Tokyo), inhibited settlement of cyprid larvae of the barnacle *Balanus amphitrite*. Bioassay-guided isolation afforded 14 active terpenoids, six of which are new. This paper describes the isolation, structural elucidation, and antifouling activity of these terpenoids.

RESULTS AND DISCUSSION

Isolation of Antifouling Compounds.

Both EtOH and MeOH extracts of the frozen marine sponge Acanthella cavernosa were separately partitioned between hexane and water. Both hexane layers showed antifouling activity against barnacle larvae. They were combined and subjected to silica gel column chromatography with a CHCl₃/MeOH system. The fraction eluted with CHCl₃ was chromatographed on a silica gel column using gradient elution from hexane to MeOH. The fraction eluted with hexane was purified by normal-phase HPLC to afford eight bioactive compounds, 1 - 8. From the more polar fractions (hexane-MeOH eluents), two bioactive compounds, 9 and 10, were obtained after purification by reverse-phase HPLC. The fractions eluted with 3-10 % MeOH-CHCl₃ from the first column were separated by reverse-phase HPLC to yield five bioactive compounds, 1 - 14.

Structures of Hydrocarbons 1, 2, and 3.

Sesquiterpene hydrocarbons 1 and 2 were identified as (+)-9-aristolene³ and (+)- α -muurolene,⁴ respectively, from spectral data including complete assignments of ¹H and ¹³C NMR signals. The absolute

10: R = NC

13: R = NCO

14: R = NCS

11: R = NC

12 : R = NHCHO

configuration of these compounds was determined by comparing $[\alpha]_D$ values with those reported in the literature.^{3,4}

Compound 3 had a molecular formula of $C_{20}H_{32}$ as determined by HREIMS. The presence of three double bonds was straightforward from the 1H and ^{13}C NMR spectra which included three olefinic proton signals (δ 5.07, 5.38, and 5.47) without any correlation to each other and six olefinic carbon signals (δ 121.5, 123.9, 125.0, 131.0, 134.4, and 136.6). Five degrees of unsaturation indicated that 3 had two rings. From comparison of NMR spectra of 3 with those of 2 it was obvious that 3 had the same ring system as 2. Extensive 2D NMR experiments disclosed that 3 was an isoprenoid unit homologue of 2 in which a methyl hydrogen of the isopropyl group of 2 was replaced by a prenyl group. Thus, 3 is biflora-4,9,15-triene, a rare biflorane diterpene. 5,6 Stereochemistry at C-11 and absolute stereochemistry of 3 remained to be elucidated.

Structures of Axane Sesquiterpenes 4 and 8.

Compound 8 was identified as (-)-cavernothiocyanate from complete assignment of ¹H and ¹³C NMR data, which were identical with those previously determined.⁷

Compounds 4 and 8 had an identical molecular formula of $C_{16}H_{25}NS$ as established by HREIMS. The presence of an isothiocyanate group in 4 was deduced from a prominent IR absorption at 2085 cm⁻¹ and a signal at δ 131.3 for the isothiocyanate carbon. From extensive NMR experiments including COSY, HMQC, and HMBC, planar structure of 4 was deduced to be 10-isothiocyanato-11-axene; all 1H and ^{13}C signals could be assigned by comparison with those of cavernothiocyanate (8). Stereochemistry of 4 was confirmed by interpretation of the NOESY spectrum. Me-14 protons (δ 0.75) showed correlation peaks with 3 β -H (δ 1.40), 7 β -H (δ 1.55), 4 β -H (δ 1.61), and 1 β -H (δ 2.9), and the angular proton (δ 0.99) at C-8 correlated with 4 α -H (δ 1.18) and 3 α -H (δ 1.25), thus revealing that the relative stereochemical assignment of 4 was 1 β -H, 7 α -Me, 8 α -H, and 9 β -Me. C-10 stereochemistry and the absolute configuration of 4 remain to be elucidated.

Structures of Aromadendrane Sesquiterpenes 5 and 7.

Compound 7 was identified as (+)-axiothiocyanate-2⁸⁻¹¹ from spectral data including complete ¹H and ¹³C NMR spectral assignments.

The molecular formula (C₁₆H₂₅NS), IR absorption (2110 cm⁻¹), and NMR data implied that compound 5 had the same planar structure as 7. Interpretation of the NOESY spectrum of 5 led to the determination of its stereochemistry, except for that at C-10. Compound 5 was one of 10-isothiocyanatoalloaromadendranes, both diastereomers of which are known.^{8,9} 1³C NMR chemical shifts of 5 were quite different from those of 10-isothiocyanatoalloaromadendrane isolated from the marine sponge *Axinella cannabina*, ⁹ but almost identical with those of 10-isothiocyanatoalloaromadendrane prepared by treatment of alloaromadendrene with HSCN.⁸ However, NOE correlations reported in the literature, ⁸ which was the only positive data to support the stereochemistry of the product by an unexpected stereoselective HSCN addition reaction, was not observed for our compound. To confirm the stereochemistry of 5 at C-10 independently from the literature, ^{8,9} we prepared both diastereomers of 10-hydroxyalloaromadendrane at C-10 non-stereospecifically from alloaromadendrene by almost the same method as reported in the literature.¹² Treatment of (-)-alloaromadendrene with *m*-chloroperbenzoic acid followed by lithium aluminum hydride reduction afforded both diastereoisomers of 10-hydroxyalloaromadendrane at C-10 in approximately a 4: 1 ratio. Comparison of their ¹H and ¹³C NMR chemical shifts and NOE data (Fig. 1) showed the major product was a 10S isomer 15 (viridiflorol) and the

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minor one (10R) 16 was ledol. From comparison of 13 C NMR chemical shifts and NOE data of 5 with those of alcohols 15 and 16, 5 was deduced to have the same stereochemistry as the 10S isomer 15, that is, 5 had the structure reported by da Silva et al.⁸ The literature reported that 1 % NOE enhancement of 6-H had been observed on irradiation of 10-Me in a 1D NOE difference experiment. Judging from our NOESY experiments, however, 1% NOE of 6-H might have resulted not from irradiation at 10-Me (δ 1.32) but that of 3α -H (δ 1.29), as the multiplet signals are very close to the 10-Me signal. As the $[\alpha]_D$ value of 5 is +8.0°, which is the opposite sign of that in the literature, δ 5 was (+)-10(R)-isothiocyanatoalloaromadendrane, the enantiomer of da Silva's compound. Hence this is the first report of compound 5 as a natural product.

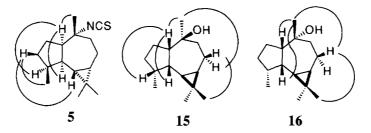


Fig. 1. NOESY correlations in 5, 15, and 16

Structures of Spiroaxane Sesquiterpenes 6 and 9.

Compound 6 was identified as (+)-axisothiocyanate-3¹³ from complete assignment of ¹H and ¹³C NMR spectral data.

From the spectral data, compound 9 was determined as axamide-3.¹³ Though the ¹H and ¹³C NMR spectra of compound 9 were observed as an equilibrium mixture of two rotational isomers (s-cis and s-trans forms) at the formamide group, ¹⁴ ¹³C NMR data of both s-cis and s-trans forms and ¹H NMR signals of the major s-trans isomer could be completely assigned from comparison of these data with those of axisothiocyanate-3 (6).

Structures of Kalihinane Diterpenes 10 - 14.

Compound 11 was identified as kalihinol-E by comparison of spectral data with those reported in the literature. 15

Compound 12 had a molecular formula of $C_{22}H_{35}CIN_2O_3$ which was established by HRFABMS. The presence of a formamide group was deduced by IR (2800 and 1670 cm⁻¹), ¹H NMR [ca. 2:3 mixture of δ 8.06 (J = 1.9 Hz) and 8.25 (12.2 Hz)], and ¹³C NMR spectra (ca. 2:3 mixture of δ 160.5 and 162.6). This was also supported by the fact that most ¹H and ¹³C NMR signals were doubled due to the presence of the equilibrium mixture (ca. 2:3) of s-cis and s-trans forms. Intense IR bands at 3300 and 2100 cm⁻¹ were reminiscent of hydroxyl and isocyanide groups, respectively. Assignments of NMR signals for each isomer (Table 1) could be accomplished by interpretation of 2D NMR spectra including HMQC and HMBC, which led to the framework of 12, identical with that of 11 (kalihinol-E). The formamide group was placed at C-10 replacing an isocyanide group in 11 from the chemical shift of C-10 (δ 59.7 for 11, and δ 55.1 and 56.4 for 12) and an HMBC correlation peak between the amide protons and C-10 for the cis-isomer. The presence of an

	10	11	12	13	14		
Assignment	s-trans s-cis		s-trans s-cis	s-trans s-cis	s-trans s-cis		
1	43.5 41.5	42.3	43.5 41.7	43.7 41.0	44.3 42.1		
2	20.9 21.3	21.7	21.0 21.4	21.2 21.6	21.0 21.0		
3	32.7 32.8	32.6	32.8 32.8	33.0 33.1	33.4 33.4		
4	70.4 70.4	70.6	70.5 70.5	71.1 71.1	71.0 71.0		
5	64.0 64.1	63.8	64.0 64.0	63.4 63.5	66.2 66.3		
6	36.7 36.5	36.1	36.7 36.5	38.4 38.2	38.6 38.1		
7	48.7 48.8	48.3	48.6 48.6	49.0 49.0	49.3 49.3		
8	22.6 22.5	21.9	22.5 22.4	22.8 22.8	23.0 22.7		
9	40.9 37.2	39.8	40.9 37.3	41.2 37.1	41.2 37.3		
10	55.0 56.4	59.8	55.1 56.4	55.1 56.6	55,1 56.5		
11	77.0 77.0	76.6	76.2 76.2	77.0 77.0	77.0 77.0		
12	38.1 38.1	28.8	28.8 28.8	38.2 38.2	38.1 38.1		
13	27.4 27.5	25.3	25.4 25.4	27.5 27.6	27.4 27.5		
14	64.2 64.4	62.9	63.0 63.0	64.3 64.5	64.2 64.4		
15	75.9 75.8	73.9	73.9 73.8	76.2 76.1	76.0 75.9		
16	22.8 22.8	28.8	28.8 28.8	23.0 23.0	23.0 23.0		
17	30.6 30.6	30.3	30.3 30.2	30.7 30.7	30.8 30.9		
18	19.2 19.1	20.5	20.5 20.5	19.4 19.3	19.4 19.3		
19	28.9 28.9	29.0	28.9 28.9	29.4 29.4	29.3 29.3		
20	19.5 19.2	20.7	19.6 19.0	19.6 19.1	19.5 19.1		
NHCHO	162.6 160.6		162.6 160.5	162.5 160.4	162.5 160.4		
NC	157.1 156.9	157.4	157.1 157.4				
NC		153.0					
NCO				122.7 120.4			
NCS					128.8 130.9		

Table 1. ¹³C NMR Data for 10, 11, 12, 13, and 14 (in CDCl₃)

Table 2. Percentages of Metamorphosed and Dead Larvae*1

% of larvae	c*2	1	2	3	4	5	6	7	8	9	10	11	12	13	14
metamorphosed	92	8	37	29	38	4	8	33	42	0	0	0	0	0	0
dead	0	0	13	0	0	0	0	0	0	0	0	0	0	0	0

^{*1 :} larvae exposed to compounds at a concentration of 5 μ g/mL for 48 h

Table 3. Percentages of Inhibition of Larval Settlement and Metamorphosis

Concentration (µg/mL)	10		12		14
5	100	100	100	100	100
0.5	69	61	49	44	67
0.05	24	24	17	50	42

^{*2 :} control

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isocyanide group at C-5 was supported from the small splitting of the 13 C signal at C-5 into a triplet by 13 C- 15 N spin-spin coupling. Finally, comparison of J_{HH} values and NOESY spectra between 11 and 12 disclosed that 12 was 10β -formamidokalihinol-E.

Spectral data of compound 10 were consistent with the same planar structure as 12 containing a formamide group. Significant differences between 10 and 12 were found in coupling constants of H-14 (δ 3.73, J = 12.3 and 4.6 Hz for the major isomer of 10; δ 3.96, J = 5.7 and 3.3 Hz for the major isomer of 12) and 13 C chemical shifts around C-14 (see Table 1). These values for 10 were quite similar to those of kalihinol-A. Extensive NMR experiments in combination with comparison with those of kalihinol-A allowed assignment of 10 as a 10-formamide analog of kalihinol-A, that is, 10β -formamidokalihinol-A.

Compound 13 had a molecular formula of $C_{22}H_{35}ClN_2O_4$, as determined by HRFABMS, containing one more oxygen atom than 10 or 12. An intense IR absorption at 2250 cm⁻¹ and a set of ^{13}C signals at δ 122.7 and 120.4 due to two rotamers suggested the presence of an isocyanate group. Most ^{13}C NMR signals of 13 were superimposable on those of 10, except for chemical shifts at C-4, C-5, and C-6 which were remarkably different from those of 10, suggesting the isocyanide group of 10 was replaced by the isocyanate group in 13. This was confirmed by 2D NMR spectra. Thus 13 was 10β -formamido-5-isocyanatokalihinol-A.

Compound 14 was deduced to be the isothiocyanate analog of 13 from the molecular formula of $C_{22}H_{35}ClN_2O_3S$ (HRFABMS), IR band (2100 cm⁻¹), and ¹³C NMR signals (δ 128.8 and 130.9; a set of signals resulted from two rotamers). Finally, extensive NMR experiments allowed assignment of 14 as 10β -formamido- 5β -isothiocyanatokalihinol-A.

Antifouling Activities against Barnacle Larvae.

Antifouling activities of isolated terpenoids against cyprid larvae of the barnacle *Balanus amphitrite* are shown in Tables 2 and 3. Table 2 summarizes percentages of metamorphosed and dead larvae within 48 hr when exposed to terpenoids 1 - 14 at a concentration of 5 µg/mL. Kalihinane derivatives 10 -14 and axamide-3 (9) inhibited larval metamorphosis completely. Table 3 shows percentages of inhibition at three concentrations for kalihinanes (10 - 14). Isocyanate and isothiocyanate derivatives 13 and 14 were highly antifouling (EC₅₀ ca. 0.05 µg/mL).

Conclusion.

A variety of terpenoids have been isolated from the marine sponge *Acanthella cavernosa* collected off Hachijo-jima Island as potential antifouling agents against cyprid larvae of the barnacle *Balanus amphitrite*. Most of them contained either isocyanate, isocyanide, or formamide group(s).

Terpenoids embracing isocyanate are very rare as natural products, especially from marine organisms. ¹⁶ It is noteworthy that kalihinol-A derivatives (10, 13, and 14) are highly antifouling.

EXPERIMENTAL

General.

IR spectra were recorded on a JASCO IR-700 spectrometer. Measurements of mass spectra were performed with a JEOL JMS-SX102A mass spectrometer. ¹H and ¹³C NMR spectra were recorded on a BRUKER ARX 500 spectrometer in CDCl₃ at 500.14 MHz and 125.77 MHz at 300 K. Chemical shifts were

reported using residual CHCl₃ (δ 7.24), CDCl₃ (δ 77.0), C₆HD₅ (δ 7.15), and C₆D₆ (δ 128.0) as internal standards. Optical rotations were measured with a JASCO DIP-140 or DIP-1000 polarimeter.

Bioassay.

Samples were dissolved in MeOH; aliquots of the solution were applied to wells of polystyrene multi-well plates and air-dried. To each well were added 2 mL of 80 % filtered sea water and six one-day-old cyprid larvae of the barnacle *Balanus amphitrite*. After incubation for 48 h in the dark at 25 °C, the numbers of larvae which attached, metamorphosed, or died were counted under a microscope.

Sponge, Extraction and Isolation of Antifoulants.

The marine sponge was collected by scuba diving at depths of $10 \sim 15$ m off Hachijo-jima Island and identified as Acanthella cavernosa Dendy, 1922 (Class Demospongiae, Order Halichondrida, Family Dictyonellidae) by Dr. Rob van Soest; a voucher specimen (ZMA POR. 6128) was deposited at the Institute for Systematics and Population Biology, University of Amsterdam. The frozen specimens (910 g) were extracted with EtOH (2×3 L) and then with MeOH (1×1 L). The EtOH extract was concentrated and partitioned between water and hexane to obtain the hexane extract (4.67 g). The MeOH extract was processed in the same way to furnish the hexane soluble material (0.48 g). Two hexane solubles were combined, and a 3.3 g portion was fractionated by silica gel column chromatography with CHCl₃, 3% MeOH/CHCl₃, and 10% MeOH/CHCl₃.

The fraction eluted with CHCl₃ (2.22 g) was chromatographed on silica gel with hexane, 50 % hexane/benzene, benzene, 50 % benzene/CHCl₃, CHCl₃, 10 % MeOH/CHCl₃, 50 % MeOH/AcOEt, and MeOH. The fraction eluted with hexane (466 mg) which showed antifouling activity was separated by normal phase HPLC with hexane to afford eight bioactive compounds (1 - 8; 1: 6.0 mg, 9.3×10^4 % yield from wet sponge; 2: 4.4 mg, 6.8×10^4 %; 3: 4.0 mg, 6.2×10^4 %; 4: 4.8 mg, 7.5×10^4 %; 5: 3.9 mg, 6.1×10^4 %; 6: 0.6 mg, 9.3 $\times 10^{15}$ %; 7: 6.1 mg, 9.5×10^4 %; 8: 3.8 mg, 5.9×10^4 %) in order of increasing polarity. Combined fraction eluted with CHCl₃ (176 mg) and 10 % MeOH/CHCl₃ (194 mg) was purified by ODS-HPLC (80 % MeOH/H₂O) to afford a bioactive formamide (9: 6.2 mg, 9.6×10^4 %), while 50 % MeOH/AcOEt (194.0 mg) was repeatedly purified by ODS-HPLC (MeOH and 80 % MeOH/H₂O) to afford a kalihinane compound (10: 7.0 mg, 1.09×10^{13} %).

The fraction eluted with 3 % MeOH/CHCl₃ (102.8 mg) from the first column was purified by ODS-HPLC (80 % MeOH/H₂O) to yield kalihinol-E (11: 3.1 mg, 4.8×10^{-4} %).

The MeOH-soluble part of the 10 % MeOH/CHCl₃ fraction (167.0 mg) from the first column was repeatedly separated by ODS-HPLC MeOH and 80 % MeOH/H₂O) to afford four kalihinane derivatives: 12 (5.8 mg, 9.0×10^{-4} %), 10 (10.0 mg, 1.6×10^{-3} %), 13 (1.9 mg, 3.0×10^{-4} %), and 14 (2.2 mg, 3.4×10^{-4} %).

1: $[\alpha]_D$ +40.0° (c 0.1, CHCl₃); ¹H NMR (CDCl₃) δ 0.60 (6-H), 0.72 (7-H), 0.91 (4-Me), 1.00 (11 β -Me), 1.02 (5-Me), 1.05 (11 α -Me), 1.22 (2 α -H), 1.30 (3 β -H), 1.42 (3 α -H), 1.58 (4-H), 1.63 (2 β -H), 1.95 (1 α -H), 2.03 (8 α -H), 2.15 (1 β -H), 2.27 (8 β -H), and 5.07 (9-H); ¹³C NMR (CDCl₃) δ 15.7 (11 α -Me), 16.0 (4-Me), 17.9 (C-11), 19.2 (C-7), 21.4 (5-Me), 21.8 (C-8), 27.2 (C-2), 29.9 (11 β -Me), 31.4 (C-3), 32.1 (C-6), 33.0 (C-1), 36.8 (C-5), 37.8 (C-4), 118.3 (C-9), and 141.6 (C-10).

2: $[\alpha]_D + 17.5^{\circ}$ (c 0.04, CHCl₃); ¹H NMR (C₆D₆) δ 0.83 (11-Me), 0.87 (11-Me), 1.48 (2-H), 1.51 (7-H), 1.66 (4-Me), 1.69 (10-Me), 1.78 (2-H and 8-H), 1.83 (3-H₂), 1.87 (8-H), 1.99 (1-H), 2.05 (11-H), 2.15 (6-H), 5.44 (9-H), and 5.56 (5-H); ¹³C NMR (C₆D₆) δ 16.1 (11-Me), 21.5 (11-Me), 21.8 (10-Me), 24.1

- (4-Me), 24.9 (C-8), 25.0 (C-2), 26.9 (C-11), 30.6 (C-3), 37.1 (C-6), 39.4 (C-1), 41.5 (C-7), 122.0 (C-9), 124.7 (C-5), 134.3 (C-4), and 136.0 (C-10).
- 3: $[\alpha]_D$ +64° (c 0.05, CHCl₃); IR (neat) v 1675 and 800 cm⁻¹; ¹H NMR (CDCl₃) δ 0.80 (12-H₃), \sim 1.2 (13-H₂), 1.35 (2-H), 1.51 (7-H), 1.57 (16Z-Me), 1.65 (16E-Me), 1.67 (10-Me), 1.68 (4-Me), 1.78 (11-H), \sim 1.8 (2-H and 8-H₂), \sim 1.9 (3-H and 14-H), 1.92 (1-H), \sim 2.0 (3-H and 14-H), 2.02 (6-H), 5.07 (15-H), 5.38 (9-H), and 5.47 (5-H); ¹³C NMR (CDCl₃) δ 13.3 (C-12), 17.6 (16Z-Me), 21.7 (10-Me), 23.9 (4-Me), 24.6 (C-2 and C-8), 25.7 (16E-Me), 26.2 (C-14), 30.8 (C-3), 31.4 (C-11), 35.7 (C-13), 36.4 (C-6), 38.9 (C-7), 39.6 (C-1), 121.5 (C-9), 123.9 (C-5), 125.0 (C-15), 131.0 (C-16), 134.4 (C-4), and 136.6 (C-10); EILRMS m/z 272 (M⁺; 22), 257 (2), and 159 (100); EIHRMS m/z 272.2510 (Calcd for C₂₀H₃₂, Δ +0.6 mmu).
- 4: $[\alpha]_D$ -52.0° (*c* 0.1, CHCl₃); IR (neat) v 2085 and 900 cm⁻¹; ¹H NMR (CDCl₃) δ 0.75 (14-H₃), 0.95 (6-H), 0.97 (15-H₃), 0.99 (8-H), 1.18 (4α-H), 1.25 (3α-H), 1.40 (3β-H), 1.48 (2-H), 1.52 (5-H₂), 1.55 (7-H), 1.61 (4β-H), 1.62 (6-H), 1.63 (2-H), 1.73 (11-Me), 2.11 (1-H), 4.47 (10-H), 4.91 and 5.06 (12-H₂); 13C NMR (CDCl₃) δ 18.3 (C-14), 20.2 (11-Me), 21.3 (C-15), 21.9 (C-5), 22.7 (C-2), 32.1 (C-7), 36.9 (C-6), 38.9 (C-4), 39.5 (C-3), 42.8 (C-1), 43.1 (C-9), 55.7 (C-8), 66.3 (C-10), 111.8 (C-12), 131.3 (-NCS), and 141.7 (C-11); EILRMS m/z 263.1697 (Calcd for C₁6H₂5NS, Δ -1.1 mmu).
- 5: $[\alpha]_D$ +8.0° (*c* 0.1, CHCl₃); IR (neat) ν 2110 cm⁻¹; ¹H NMR (CDCl₃) δ 0.11 (6-H), 0.63 (7-H), 0.92 (4-Me), 1.017 (11α-Me), 1.022 (11β-Me), 1.29 (3β-H), 1.32 (10-Me), 1.34 (8α-H), 1.54 (2β-H), 1.61 (2α-H), 1.72 (9-H₂), 1.80 (8β-H), 1.81 (3α-H), 1.87 (5-H), 2.02 (4-H), and 2.10 (1-H); ¹³C NMR (CDCl₃) δ 16.1 (4-Me and 11α-Me), 18.7 (C-11), 20.6 (C-8), 22.2 (C-6), 24.4 (C-2), 28.2 (C-7), 28.6 (11β-Me), 28.9 (C-3), 30.4 (10-Me), 37.2 (C-9), 38.1 (C-4), 40.6 (C-5), 57.3 (C-1), and 68.4 (C-10); EILRMS m/z 263 (M⁺; 100), 248 (21), 205 (3), and 121 (58); EIHRMS m/z 263.1721 (Calcd for C₁₆H₂₅NS, Δ +1.3 mmu).
- **6**: $[\alpha]_D$ +52.9° (c 0.07, CHCl₃); 1 H NMR (CDCl₃) δ 0.75 (10-Me), 0.89 (11-Me), 0.91 (11-Me), 1.08 (9α-H), 1.22 (7-H), 1.25 (8β-H), 1.49 (11-H), 1.52 (9β-H), 1.67 (10-H), 1.72 (3-Me), 1.78 (8α-H), 1.87 (1β-H), 1.93 (1α-H), 2.22 (2-H₂), 3.67 (6-H), and 5.12 (4-H); 13 C NMR (CDCl₃) δ 16.1 (10-Me), 16.9 (3-Me), 20.4 (11-Me), 20.8 (11-Me), 25.4 (C-8), 30.2 (C-11), 31.3 (C-9), 35.0 (C-10), 35.1 (C-1), 35.9 (C-2), 45.4 (C-7), 59.0 (C-5), 67.4 (C-6), 124.0 (C-4), and 144.8 (C-3).
- 7: $[\alpha]_D$ +75.0° (c 0.1, CHCl₃); IR (neat) v 2085 cm⁻¹; 1H NMR (CDCl₃) δ 0.53 (6-H), 0.59 (7-H), 0.90 (4-Me), 0.94 (8 α -H), 0.95 (11 α -Me), 0.99 (11 β -H), 1.25 (10-Me), 1.26 (3 β -H), 1.26 (5-H), 1.42 (2 α -H), 1.68 (3 α -H), 1.80 (8 β -H), 1.82 (9 β -H), 1.87 (2 β -H), 1.92 (9 α -H), 2.02 (4-H), and 2.18 (1-H); 13 C NMR (CDCl₃) δ 15.8 (11 α -Me), 16.0 (4-Me), 19.3 (10-Me), 19.8 (C-8), 20.4 (C-11), 26.6 (C-7), 27.5 (C-2), 28.4 (C-6), 28.5 (11 β -Me), 34.3 (C-3), 36.7 (C-4), 39.2 (C-5), 43.3 (C-9), 56.6 (C-1), 66.6 (C-10), and 129.3 (-NCS).
- 8: $[\alpha]_D$ -6.0° (c 0.1, CHCl₃); 1H NMR (CDCl₃) δ 0.75 (8-H), 0.77 (15-H₃), 0.78 (14-H₃), 0.81 (6-H), 1.09 (4-H), 1.24 (3-H), 1.25 (2-H), 1.42 (3-H), 1.44 (7-H), 1.48 (5-H₂), 1.61 (6-H), 1.62 (4-H), 1.75 (11-Me), 2.01 (2-H), 2.51 (1-H), 3.47 and 3.55 (12-H₂), and 5.57 (11-H); ^{13}C NMR (CDCl₃) δ 14.9 (11-Me), 18.5 (C-14), 21.3 (C-15), 22.2 (C-5), 29.8 (C-2), 32.3 (C-7), 36.8 (C-6), 39.4 (C-4), 39.7 (C-1), 39.8 (C-3), 42.8 (C-9), 44.5 (C-12), 60.7 (C-8), 124.5 (C-11), and 141.9 (C-10).
- 9: $[\alpha]_D$ -150 (c 0.06, CHCl₃); ¹H NMR of major s-trans isomer (CDCl₃) δ 0.75 (10-Me), 0.86 (11-Me), 0.90 (11-Me), 0.95 (8-H), \sim 1.1 (9-H), 1.25 (11-H), 1.28 (7-H), 1.38 (10-H), \sim 1.5 (9-H), \sim 1.6 (1-H), \sim 1.7

 $(3-Me), \sim 1.8 \ (8-H), \ 1.83 \ (1-H), \ 2.05 \sim 2.35 \ (2-H_2), \ 4.24 \ (6-H), \ 5.23 \ (4-H), \ and \ 8.24 \ (-NHCHO); \ ^{13}C \ NMR \ of major s-\textit{trans} \ isomer \ (CDCl_3) \ \delta \ 16.2 \ (10-Me), \ 17.0 \ (3-Me), \ 20.9 \ (11-Me), \ 21.2 \ (11-Me), \ 25.3 \ (C-8), \ 29.5 \ (C-11), \ 31.6 \ (C-9), \ 33.6 \ (C-1), \ 35.6 \ (C-10), \ 36.0 \ (C-2), \ 44.2 \ (C-7), \ 58.1 \ (C-5), \ 53.9 \ (C-6), \ 125.0 \ (C-4), \ 143.6 \ (C-3), \ and \ 160.9 \ (-NHCHO).$

10: $[\alpha]_D$ +11.0° (*c* 0.145, CHCl₃); IR (neat) v 3300, 2150, and 1670 cm⁻¹; ¹H NMR of major s-*trans* isomer (CDCl₃) δ 1.09 (8-H), 1.17 (11-Me), 1.21 (10-Me), 1.32 (15α-Me), 1.33 (15β-Me), 1.38 (4-Me), 1.44 (1-H), 1.45 (2-H), 1.52 (12-H), 1.54 (3-H), 1.60 (9-H and 12-H), 1.61 (2-H), 1.65 (7-H and 8-H), 1.76 (3-H), 1.81 (9-H), 2.00 (13-H), 2.06 (13-H), 2.10 (6-H), 3.73 (dd, 12.3 and 4.6 Hz, 14-H), 4.54 (5-H), 5.52 (NH), and 8.25 (CHO); ¹³C NMR see Table 1; FABLRMS (glycerol, positive) m/z 503, 505 (M+glycerol+H)⁺, 411, 413 (M+H)⁺, and 366; FABHRMS m/z 411.2426 (Calcd for C₂₂H₃₅³⁵ClNO₃+H, Δ +1.2 mmu).

11: $[\alpha]_D + 4.2^{\circ}$ (c 0.09, CHCl₃); ¹³C NMR see Table 1.

12: $[\alpha]_D$ -2.80 (*c* 0.145, CHCl₃); IR (neat) v 3300, 2800, 2100, 1670, and 750 cm⁻¹; ¹H NMR of major s-trans isomer (CDCl₃) δ 1.08 (8-H), 1.17 (11-Me), 1.22 (10-Me), 1.31 (12-H), 1.33 (15α-Me), 1.34 (15β-Me), 1.39 (4-Me), 1.48 (1-H), 1.53 (3-H), 1.63 (9-H), 1.77 (8-H), 1.78 (3-H and 7-H), 1.81 (2-H), 1.82 (9-H), 1.94 (13-H), 1.95 (12-H), 2.10 (6-H), 2.32 (13-H), 3.96 (dd, 5.7 and 3.3 Hz, 14-H), 4.60 (5-H), 5.62 (NH), and 8.25 (CHO); ¹³C NMR see Table 1; FABLRMS (glycerol, positive) m/z 503, 505 (M+glycerol+H)⁺, 411, 413 (M+H)⁺, and 366; FABHRMS m/z 411.2417 (Calcd for C₂₂H₃₅³⁵ClNO₃+H, Δ +0.3 mmu).

13: $[\alpha]_D$ -25° (*c* 0.02, CHCl₃); IR (neat) v 3300, 2250, 1670, and 700 cm⁻¹; ¹H NMR of major s-trans isomer (CDCl₃) δ 1.09 (8-H), 1.20 (10-Me), 1.23 (11-Me), 1.27 (4-Me), 1.33 (15α-Me), 1.36 (15β-Me), 1.39 (1-H), 1.45 (3-H), 1.46 (9-H), 1.50 (7-H), 1.53 (12-H), 1.59 (12-H), 1.64 (3-H), 1.67 (8-H), 1.81 (9-H), 2.00 (13-H), 2.06 (13-H), 2.13 (6-H), 3.74 (14-H), 4.47 (5-H), 5.50 (NH), and 8.24 (CHO); ¹³C NMR see Table 1; FABLRMS (glycerol, positive) m/z 519, 521 (M+glycerol+H)⁺, 427, and 429 (M+H)⁺; FABHRMS m/z 427.2361 (Calcd for C₂₂H₃₅³⁵ClNO₄+H, Δ -0.3 mmu).

14: $[\alpha]_D + 24^0$ (c 0.05, CHCl₃); IR (neat) v 3300, 2100, 1630, and 770 cm⁻¹; 1H NMR of major s-*trans* isomer (CDCl₃) δ 1.07 (8-H), 1.20 (11-Me), 1.21 (10-Me), 1.33 (4-Me), 1.34 (15-Me x 2), 1.35 (1-H), 1.54 (12-H), 1.55 (3-H), 1.59 (7-H), 1.58 (9-H), 1.61 (12-H), 1.65 (3-H), 1.67 (8-H), 1.81 (9-H), 2.01 (13-H), 2.06 (13-H), 2.14 (6-H), 3.75 (14-H), 4.67 (5-H), 5.50 (NH), and 8.26 (CHO); ^{13}C NMR see Table 1; FABLRMS (glycerol, positive) m/z 443, 445 (M+H)⁺, 398, 400, 380, and 382; FABHRMS m/z 443.2138 (Calcd for $C_{22}H_{35}^{35}ClNO_3S+H$, Δ +0.3 mmu).

Preparation of 10-hydroxyalloaromadendranes (15 and 16).

To a solution of (-)-alloaromadendrane (43 mg; commercially available from Fluka) in chloroform (4 mL) was added *m*-chloroperbenzoic acid (70 %; 63 mg), and the mixture was stirred at rt for 30 min. After addition of aqueous sodium bisulfite solution followed by sodium bicarbonate solution, the mixture was extracted with chloroform and washed with H₂O. After evaporation of chloroform, the residue (48 mg) was dissolved in ether (5 mL), to which was added a large excess of lithium aluminum hydride (10 mg), and the mixture was stirred at rt overnight. After addition of 2 M HCl, the reaction mixture was extracted with ether; the ether extract was successively washed with brine, sodium bicarbonate solution, and 2 times with brine; then dried over sodium sulfate, affording 49 mg of reaction product mixture after evaporation. The mixture was passed through a silica gel (13 g) column with chloroform to yield a mixture of 10-hydroxyalloaromadendrane (15 and 16; ca. 30 mg),

which was purified by HPLC (SiO_2 ; CHCl₃) to afford (+)-10(S)-hydroxyalloaromadendrane (15; 15 mg) and (-)-10(R)-hydroxyalloaromadendrane (16; 4 mg).

15: [α]_D +2.1° (c 0.1, CHCl₃); 1 H NMR (CDCl₃) δ 0.09 (6-H), 0.59 (7-H), 0.90 (4-Me), 0.97 (11β-Me), 1.02 (11α-Me), 1.13 (10-Me), 1.24 (3α-H), 1.37 (8β-H), 1.54 (9α-H), 1.53 (2α-H), 1.62 (8α-H), 1.63 (2β-H), 1.69 (9β-H), 1.78 (3β-H), 1.80 (1-H), 1.81 (5-H), and 1.95 (4-H); 13 C NMR (CDCl₃) δ 16.1 (11β-Me), 16.3 (4-Me), 18.4 (C-11), 18.8 (C-8), 22.3 (C-6), 25.8 (C-2), 28.6 (C-7), 28.7 (11α-Me), 29.1 (C-3), 32.1 (10-Me), 37.8 (C-9), 38.5 (C-4), 39.7 (C-5), 58.2 (C-1), and 74.6 (C-10); EILRMS m/z 222 (M⁺; 20), 204 (100), and 189 (70); EIHRMS m/z 222.1968 (Calcd for C₁₅H₂₆O, Δ -1.6 mmu). 16: [α]_D -10.5° (c 0.1, CHCl₃); 1 H NMR (CDCl₃) δ 0.31 (6-H), 0.70 (7-H), 0.91 (4-Me), 0.98 (11β-Me), 1.02 (11α-Me), 1.12 (10-Me), 1.19 (8β-H), 1.28 (3α-H), 1.66 (2β-H), 1.67 (9-H), 1.68 (3β-H), 1.76 (5-H), 1.80 (8α-H), 1.83 (9-H), 1.87 (2α-H), 1.97 (4-H), and 2.07 (1-H); 13 C NMR (CDCl₃) δ 15.4 (11β-Me), 16.0 (4-Me), 19.2 (C-11), 20.3 (C-8), 23.4 (C-6), 24.6 (C-2), 25.0 (C-7), 28.6 (11α-Me), 30.5 (10-Me), 30.8 (C-3), 38.4 (C-4), 39.2 (C-9), 40.8 (C-5), 53.8 (C-1), and 74.6 (C-10); EILRMS m/z 222 (M⁺; 78), 204 (100), and 189 (92); EIHRMS m/z 222.2000 (Calcd for C₁₅H₂₆O, Δ +1.7 mmu).

Acknowledgment: We thank Professor Isao Kitagawa, Kinki University, for his valuable suggestions, and Professor Paul J. Scheuer, University of Hawaii, for reading the manuscript. Thanks are also due to Dr. Rob van Soest, University of Amsterdam, for sponge identification.

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